

Fission

- _____ of a nucleus
- Releases a lot of _____
- An unstable nucleus can naturally decay with α or β radiation, but can take a long time
- _____ done by hitting a large nucleus with a _____ (β radiation)

Chain reaction

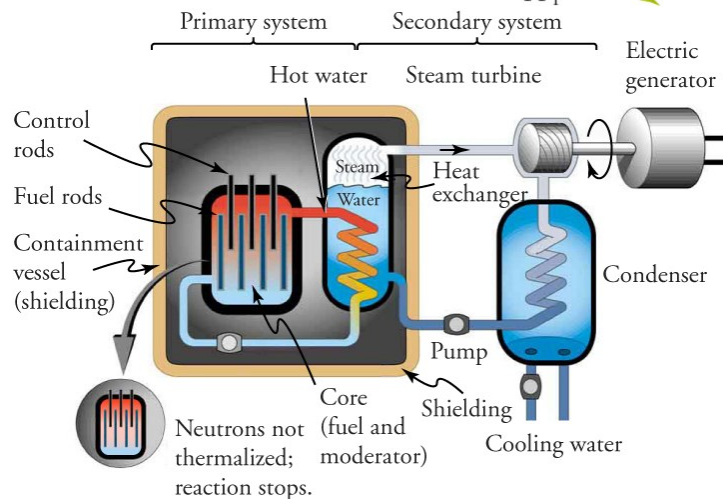
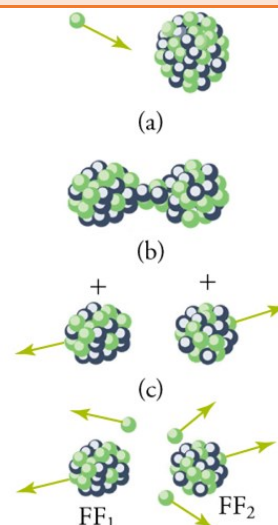
- When the nucleus splits it releases free _____
- Those can _____ other nuclei and _____ them
- Critical mass – Minimum amount of _____ material necessary to sustain fission _____ reaction
- Number of fission reactions increases _____

Nuclear Reactor

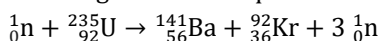
- To keep a nuclear fission reaction from becoming a _____, slow down the neutrons with _____
- Fuel rods contain _____
- Control rods _____ neutrons
 - Insert control rods to _____ reaction
- Fission reaction _____ water
- Steam turns turbines to make _____
- _____ water goes back to be heated

Energy from Fission

- The mass of the products of fission is _____ than parent nucleus
- That mass is converted to _____ by $E = mc^2$
- Average fission reaction produces about _____ MeV of energy



Find the energy released in the fission of uranium-235 given in the equation



Neutron: 1.008665 u, ${}^{235}\text{U}$: 235.0439299 u, ${}^{141}\text{Ba}$: 140.9144035 u, ${}^{92}\text{Kr}$: 91.926173094 u

Calculate the amount of energy produced by the fission of 1.00 kg of ^{239}Pu , given the average fission reaction of ^{239}Pu produces 211.5 MeV. The atomic mass of ^{239}Pu is 239.05 u.

Practice Work

- How can a nuclear reactor contain many critical masses and not go supercritical? What methods are used to control the fission in the reactor? (OpenStax C32.23)
- If a nucleus elongates due to a neutron strike, which of the following forces will decrease? (HSP 22.20)
 - Nuclear force between neutrons only
 - Coulomb force between protons only
 - Strong nuclear force between all nucleons and Coulomb force between protons, but the strong force will decrease more
 - Strong nuclear force between neutrons and Coulomb force between protons, but Coulomb force will decrease more
- (a) Calculate the energy released in the neutron-induced fission (similar to the spontaneous fission in Example 32.3) $n + {}^{238}_{92}\text{U} \rightarrow {}^{96}_{38}\text{Sr} + {}^{140}_{54}\text{Xe} + 3n$, given $m({}^{238}_{92}\text{U}) = 238.050783 \text{ u}$, $m({}^{96}_{38}\text{Sr}) = 95.921750 \text{ u}$ and $m({}^{140}_{54}\text{Xe}) = 139.92164 \text{ u}$. (b) This result is about 6 MeV greater than the result for spontaneous fission. Why? (c) Confirm that the total number of nucleons and total charge are conserved in this reaction. (OpenStax 32.43) **177.0 MeV; 239 nucleons, 92 + charges**
- (a) Calculate the energy released in the neutron-induced fission reaction $n + {}^{235}_{92}\text{U} \rightarrow {}^{92}_{36}\text{Kr} + {}^{142}_{56}\text{Ba} + 2n$, given $m({}^{235}_{92}\text{U}) = 235.043923 \text{ u}$, $m({}^{92}_{36}\text{Kr}) = 91.926269 \text{ u}$ and $m({}^{142}_{56}\text{Ba}) = 141.916361 \text{ u}$. (b) Confirm that the total number of nucleons and total charge are conserved in this reaction. (OpenStax 32.44) **179.4 MeV; 236 nucleons, 92 + charges**
- (a) Calculate the energy released in the neutron-induced fission reaction $n + {}^{239}_{94}\text{Pu} \rightarrow {}^{96}_{38}\text{Sr} + {}^{140}_{56}\text{Ba} + 4n$, given $m({}^{239}_{94}\text{Pu}) = 239.0521634 \text{ u}$, $m({}^{96}_{38}\text{Sr}) = 95.921750 \text{ u}$ and $m({}^{140}_{56}\text{Ba}) = 139.910581 \text{ u}$. (b) Confirm that the total number of nucleons and total charge are conserved in this reaction. (OpenStax 32.45) **180.6 MeV; 240 nucleons, 94 + charges**
- The naturally occurring radioactive isotope ${}^{232}_{90}\text{Th}$ does not make good fission fuel, because it has an even number of neutrons; however, it can be bred into a suitable fuel (much as is bred into ${}^{239}_{92}\text{U}$). (a) What are Z and N for ${}^{232}_{90}\text{Th}$? (b) Write the reaction equation for neutron captured by ${}^{232}_{90}\text{Th}$ and identify the nuclide ${}^A_Z\text{X}$ produced in $n + {}^{232}_{90}\text{Th} \rightarrow {}^A_Z\text{X} + \gamma$. (c) The product nucleus β^- decays, as does its daughter. Write the decay equations for each, and identify the final nucleus. (d) Confirm that the final nucleus has an odd number of neutrons, making it a better fission fuel. (e) Look up the half-life of the final nucleus to see if it lives long enough to be a useful fuel. (OpenStax 23.48) **$Z = 90$, $N = 142$; Thorium; Daughters are ${}^{233}_{91}\text{Pa}$ and ${}^{233}_{92}\text{U}$; 141 neutrons; 160000 yrs**
- The electrical power output of a large nuclear reactor facility is 900 MW. It has a 35.0% efficiency in converting nuclear power to electrical. (a) What is the thermal nuclear power output in megawatts? (b) How many ${}^{235}_{92}\text{U}$ nuclei fission each second, assuming the average fission produces 200 MeV? (c) What mass of ${}^{235}_{92}\text{U}$ is fissioned in one year of full-power operation? (OpenStax 32.49) **2570 MW; 8.04×10^{19} fissions/s; 990 kg**